

Production of Natural Biopharmaceuticals from the Microalgae Living in the Dead Sea

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Abstract

This study investigates the feasibility of β -carotene production from *Dunaliella* isolated from the Dead Sea employing a number of interdependent steps and focusing on the laboratory scale cultures. Then the produced Beta Carotene was subjected to enzymatic oxidation to produce tretinoin (vitamin A). The produced vitamin A was verified using proper analysis. *Dunaliella* was isolated from the Dead Sea and cultivated using a certain media until the cell count was 500,000 cell/ml. The cultivation step was monitored and video captured using a digital microscope with USB camera, then it was centrifuged and extracted using organic solvents and oils like jojoba. The amount of dry weight was 2.1 g/L of which 3-6% was Beta Carotene. Freeze drying step was performed to obtain Beta Carotene as powder. The analysis of the powder was carried out by the research and development department at one of the local biopharmaceutical companies. It was evident that pilot plant investigations should be the next step on the way of commercialization of such a profitable process.

Key Words: Biopharmaceuticals, Microalgae, Beta Carotene, Cultivation, *Dunaliella*, Vitamin A.

1. Introduction

1.1 General:

The unicellular biflagellate algae "*Dunaliella*" is the richest natural source of the carotenoid β -carotene. β -carotene is generally regarded as the most commercially important and widely used carotenoid. It is used as a food coloring agent, an antioxidant and an important and safe pro-vitamin A source. Nowadays there is a global tendency to develop a cancer fighter medicine based on the antioxidation feature of β -carotene. The halophilic species of *Dunaliella* also accumulates high concentrations of glycerol. *Dunaliella* was first proposed as a commercial source of β -carotene and latter as a source of glycerol (Chidambara, 2005).

β -carotene from *Dunaliella* is now being produced on a commercial scale in Australia, the USA and Israel. (Oren, 1999).

1.2 Description of *Dunaliella*

What is *Dunaliella*?

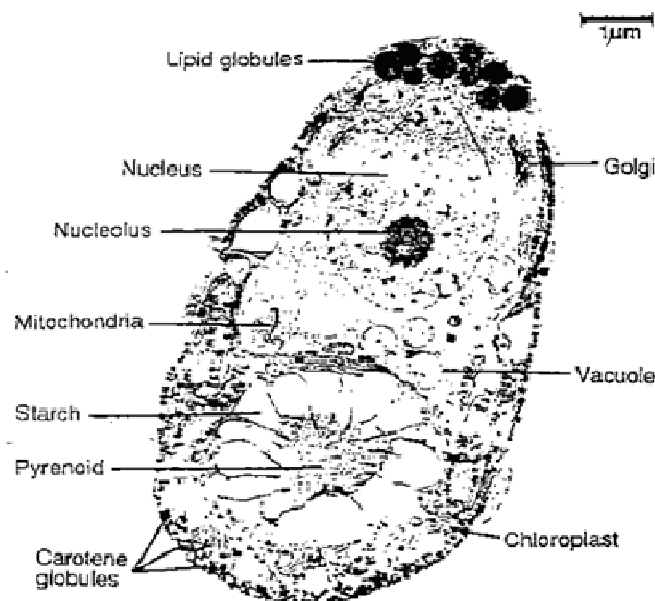


Fig (1.2.1): A photo of *Dunaliella* cell, showing its main organs.

It is a unicellular, bi-flagellate, naked green alga which is morphologically similar to chlamydomonas with the main difference being the absence of a cell wall in Dunaliella.

It has two flagella of equal length and a single cup-shaped chloroplast. In *Dunaliella Salina* and *Dunaliella Parva* the chloroplast accumulates large quantities of β -carotene (as droplets) so that the cells appear orange-red rather than green. (Avron et al, 1999). The first description of the genus was given by Teodoresco who reported occurrence of the organism we know today as *Dunaliella*. At present a total of 29 species as well as a number of varieties and forms are recognized. Cell shape in this genus is very variable, being oval, spherical, cylindrical, ellipsoidal, egg-pear-or spindle shaped. Cell in any given species may change shape with changing conditions, often coming spherical under unfavorable conditions. Cell size also varies with growth conditions and light intensity (Ben-Amotz et al, 2012).

Dunaliella species are commonly observed in salt lakes in all parts of the world from tropical to temperate to polar regions where they often impart an orange-red color to the water.

When exposed to stress conditions such as high light intensity or nutrient starvation, two stereoisomers of β -carotene, all-**trans** and 9-**cis** β -carotene, accumulated, reaching up to 14% of the cell's dry weight, with the pigment being deposited into plastid. (Emeish, 2012)

We now know that not all *Dunaliella* species produce massive amounts of carotene and those that can do so only under suitable conditions. The following figures illustrates some of *Dunaliella* types.

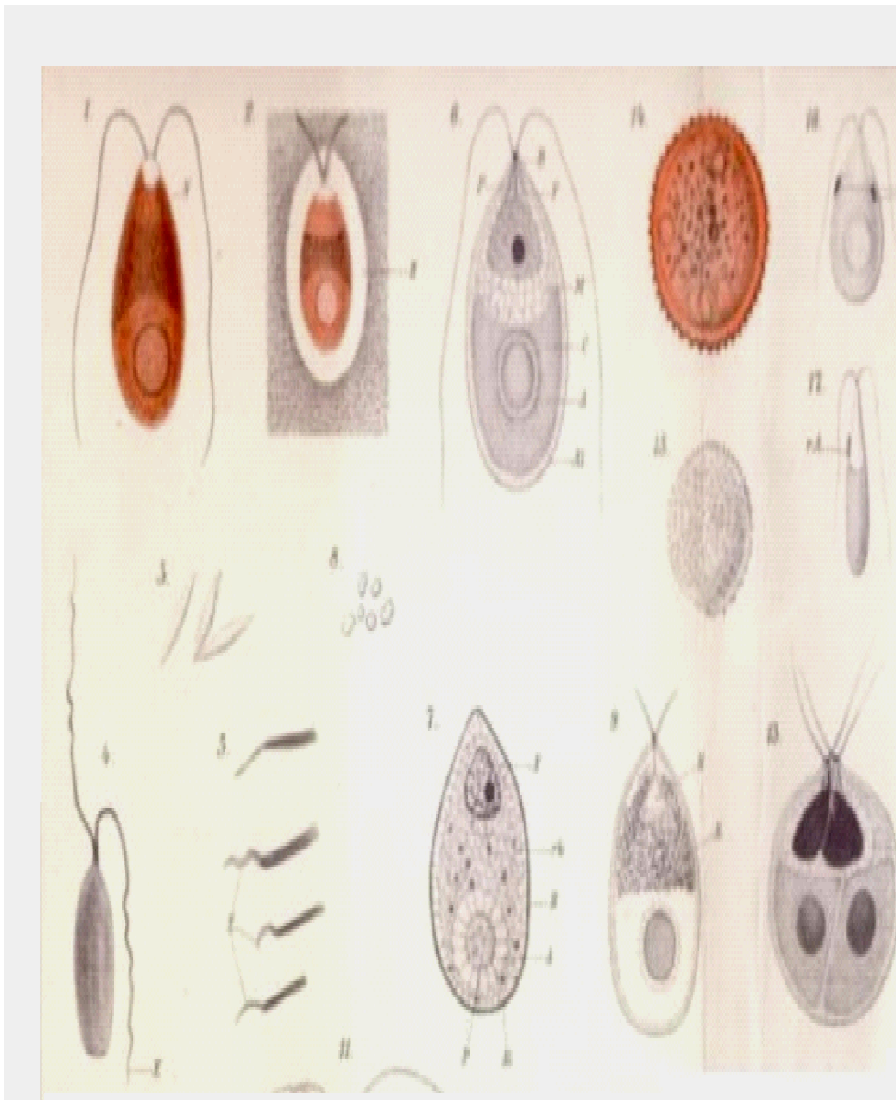


Fig. (1.2.3): A photo of different *Dunaliella* types.

1.3 β -carotene

1.3.1 Structure:

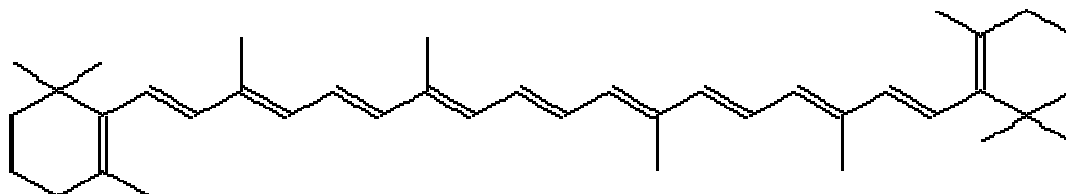


Fig. (1.3.1): β -carotene structure $C_{40}H_{56}$

Its chemical composition is:

1,1'-(3,7,12,16-Tetramethyl-1,3,5,7,9,11,13,15,17-octadeca-nonaene-1,18-diyl)bis[2,6,6-trimethylcyclohexane].

And its molecular weight is 536.87 g.

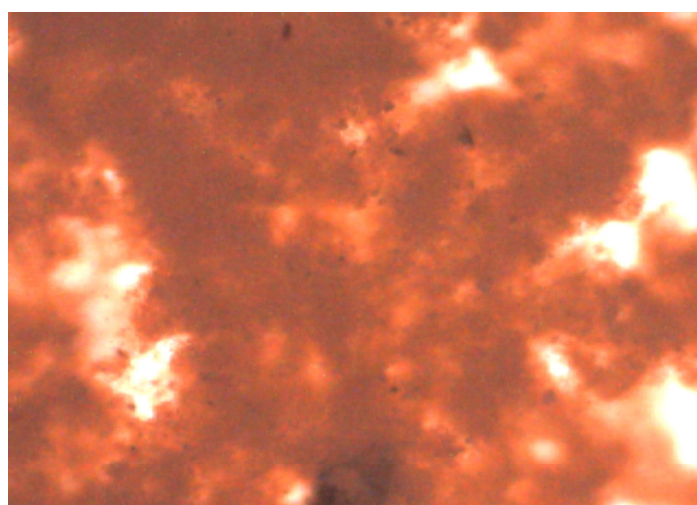


Fig.(1.3.2)Produced Beta Carotene photo under microscope

The pigment responsible for the brightly red coloration displayed by *Dunaliella* was recognized already very early as a carotenoid. (Abusara et al, 2011).It can be found in carrot, orange, and many other yellow, orange or red colored fruits and vegetables. It also accumulates in large quantities in the chloroplast of *Dunaliella* cells as droplets so that the cells appear orange-red rather than green.

A typical composition, specialize the *Dunaliella* β -carotene from other sources, expressed as percentage of total β -carotene is: 10% 15-cis- β -carotene, 41% 9-cis- β -carotene, 42% all-trans- β -carotene, and 6% other isomers. (Avron et al, 1997)

In the alga, this β -carotene seems to act a sun screen to protect the chlorophyll and the cell DNA from high irradiance, which characterizes the normal habitat of *Dunaliella*.

1.3.2 Biosynthesis of β -carotene:

The general biosynthesis of β -carotene can be divided into four stages:

1. Formation of Geranylgeranyl pyrophosphate (GGDP) from Mevalonic acid.
2. Condensation to form phytoene.
3. Desaturation of phytoene to lycopene.
4. Cyclization of lycopene to form β -carotene.

The unique caroteneogenesis in *Dunaliella* has been used to study the last two stages, (3) and (4), by applying rapid solvent extraction.

1.4 Literature review

A hundred years have passed since the description of the genus *Dunaliella*, the unicellular alga which is responsible for most of the primary production of various types of compounds and phenomena in hypersaline environments worldwide as salt mushroom formation phenomena within the Dead Sea solar ponds.(Daghistani et al, 2003).

First sighted in 1838 in saltern evaporation ponds in south France by Michel Flix Dunal who reported occurrence of the organism we know today as *Dunaliella Salina* in the salterns of Montpellier, on the coast of France.

In the century that has elapsed since its formal description the genus *Dunaliella* has become a convenient model organism for the study of salt adaptation in alga. The establishment of the concept of organic compatible solutes to provide osmotic balance was largely based on the study of *Dunaliella* species. Moreover, the massive accumulation of β -carotene by some strains under suitable growth conditions has led to interesting biotechnological applications.

In the course of 19th century, Dunal's red flagellate alga has been observed by other biologists as well in salt lakes and hypersaline sites in many countries. Different names were attached to the organism by each investigator. (Emeish et al, 2003).

It was named after its discoverer by Teodoresco in 1905, this year saw the publication of two papers presenting in-depth descriptions of *Dunaliella* as anew genus one by Teodoresco and the second by Carla Hamburger.

Teodoresco studied materials collected from a Romanian salt lake, while Hamburger worked with samples sent to her from salterns of Sardinia. Both authors presented detailed drawings of the organisms and provided extensive information on its morphology, cell structure, reproduction, behavior and ecology.

A formal description of the genus *Dunaliella*, observed by Dunal who had first seen these organisms in salterns in France almost seventy years earlier was published in 1906.

In the Dead Sea, *Dunaliella* cells have been reported since 1940s.

The first quantitative estimates of the *Dunaliella* population in the lake were made in 1964 and showed very high numbers up to 4×10^4 cells per ml of surface water.

Stephens and Gillespie in 1976 reported measurements of the primary production in the south arm of Great Salt Lake performed in 1973.

An in-depth taxonomic treatment of the genus was given in Massyuk's 1973 monograph.

Dead Sea from 1980 onwards has yielded a clear picture of the factors that determine development of the alga in this unusual environment.

A Master thesis under the title of:

"Optimal Growth Conditions for the Production of β -Carotene and Glycerol from a Halophilic Alga *Dunaliella* Species Isolated from the Dead Sea" was presented In January, 2004 by Nader Abu Sara from Jordan University of Science and Technology .(Abusara,2004).

In that research *Dunaliella* species were isolated from scattered ponds of the southern part of the Dead Sea and the conditions that gave the highest growth and β -carotene production were studied.

Also the effects of different physical (temperature and light intensity) and chemical (different nitrogenous and sulfate compounds) factors on the growth and carotenoids production rates were reported. (Wheater, 2007).

With a study published in 2004 by Liska, *Dunaliella* research has entered the era of modern proteomics.

That was a worthy conclusion of the first century of *Dunaliella* research, and provides us with a preview of the kind of information that may be obtained in the coming years.

1.5 The Dead Sea

1.5.1 Ecology:

The Dead Sea is a terminal lake in the Syrian-African rift valley with a rhomb-shaped.

The Dead Sea water level (425 m below mean sea water is the lowest exposed surface on earth; which is 76 km long, up to 18 km wide and 400 m deep, while its bottom (825 m below mean seawater level) forms the lowest continental surface.

The Dead Sea has a unique Ca-chloride composition [i.e., $\text{Ca}^{2+} > (\text{SO}_4^{2-} + \text{HCO}_3^-)$] forming one of the world's saltiest lake with a total dissolved salt concentration of 340 g/L and a density of about 1.235 g/ml. At that time, the lake was stratified and consisted of 2 basins: a deep one on the north and a shallow one in the south, in 1976, the southern basin were detached and today the area is occupied by a series of evaporation ponds that are filled by pumping water from the northern basin.(Haugen, 1986).The rise and decline of a bloom of the algae in the Dead Sea is well observed. In the beginning of 1980 a massive amount of fresh water from the Jordan river and rain floods in flowed to the lake causes a sufficient dilution of the upper water layers, and as a consequence a dense *Dunaliella* bloom developed; the algal bloom remained present for a few months only.During the winter of 1991-1992 another massive amounts of rain water being transported to the Dead Sea, which caused an increase in level of almost 2 m. A dilution of the upper 5 meters to a salinity as low as 245 g / L triggered a short-lived mass development of *Dunaliella* and a prolonged bloom of red algae.(Emeish et al, 2003)

The negative water balance of the Dead Sea since 1930 to 1999 has resulted in a decrease in its water level by about 20 m, this has been a companied by an increase in the salinity of the brine and by various change in the chemical and physical characteristics of the lake.

1.5.2 *Dunaliella* in the Dead Sea:

The eukaryotic alga *Dunaliella* was first reported to be present in the Dead Sea in 1940 by Elzari-Volcani. (Oren.1999).

Twenty different species of microalgae and halophilic bacteria were isolated and identified from eastern shores

of the Dead Sea.

Multispectral LANDSAT images of the Dead Sea area were analyzed to obtain spatial and temporal information on the development of the algae bloom.

Two images were obtained: the first, on April 15 1992 at the beginning of Dunaliella bloom, and the second, on June 22 from the same year, when the algal bloom had declined and the Dead Sea appeared red. For comparison, a LANDSAT image was also obtained on May, 5, 1991, at the time when algae were absent. The spectral distribution of the reflected solar radiation is modified by pigments present in the micro algae. The main interaction commonly measured is the absorption of sunlight by the red band of chlorophyll contained in the algal plastid.

2. Methodology:

2.1 Cultivation:

The microalgae was cultivated in a 5 liters vessels exposed to neon lamp from upper and lower sides at room temperature (about 25°C) which kept constant during the different seasons. Ordinary air was supplied.

Modified Johnson medium (table 2.1.1) was used as nutrient source for algae cultivation. The cultivation of Dunaliella was monitored by using a digital microscope with USB camera connected to computer to follow the algal cultivation under different light intensities .

Table 2.1.1: Modified Johnson Medium (Borowitzka, 1990)

To 5 L of distilled water add :	
NaCl	24% per unit volume
MgCl ₂ .6H ₂ O	7.65g
MgSO ₄ .7H ₂ O	2.6g
KCl	1.02g
CaCl ₂ .2H ₂ O	1.02g
KNO ₃	5.1g
NaHCO ₃	0.2g
KH ₂ PO ₄	0.2g
Fe-solution	51ml
Trace- element solution	51ml
Na ₂ EDTA	945mg
FeCl ₃ .6H ₂ O	1220mg
H ₃ BO ₃	305mg
(NH ₄) ₆ Mo ₇ O ₂₄ .4H ₂ O	190mg
CuSO ₄ .5H ₂ O	30mg
CoCl ₂ .6H ₂ O	25.5mg
ZnCl ₂	20.5mg
MnCl ₂ .4H ₂ O	20.5mg
Adjust PH to 7.5 with HCl	

2.2 Cell Counting:

To be sure that the cells are ready to be harvested, cell counting step was undertaken. Counting of Dunaliella cells was performed using a Hemocytometer slide of 0.1 mm depth. No sample dilution has been done, 100 µl of the culture was taken and tested to count the cell number, and the trial was repeated 3 times.

A simple filtration followed by long spontaneous drying in dark was used to approximate the algal cell density.

2.3 Harvesting:

After the cultivation media reaches to a suitable cells density it was isolated from the cultivation media and transferred into smaller vessels.

2.4 Centrifugation:

Then 25 ml culture media was centrifuged at 4000 rpm for 5 min. The sample was re suspended in 5ml distilled water.

Continues flow centrifuge of type HERMLE z300 rotates by 4000 rpm was employed to perform the mentioned.

2.5 Extraction:

β-carotene was extracted with hexane, ether, jojoba oil or light vegetable oil separately by adding 25 ml cell suspension to 15 ml solvent. When the extraction was performed, oil was heated to 70°C; after it was cooled down to 40° C, then β-carotene was added. (See Fig.(2.5.1))

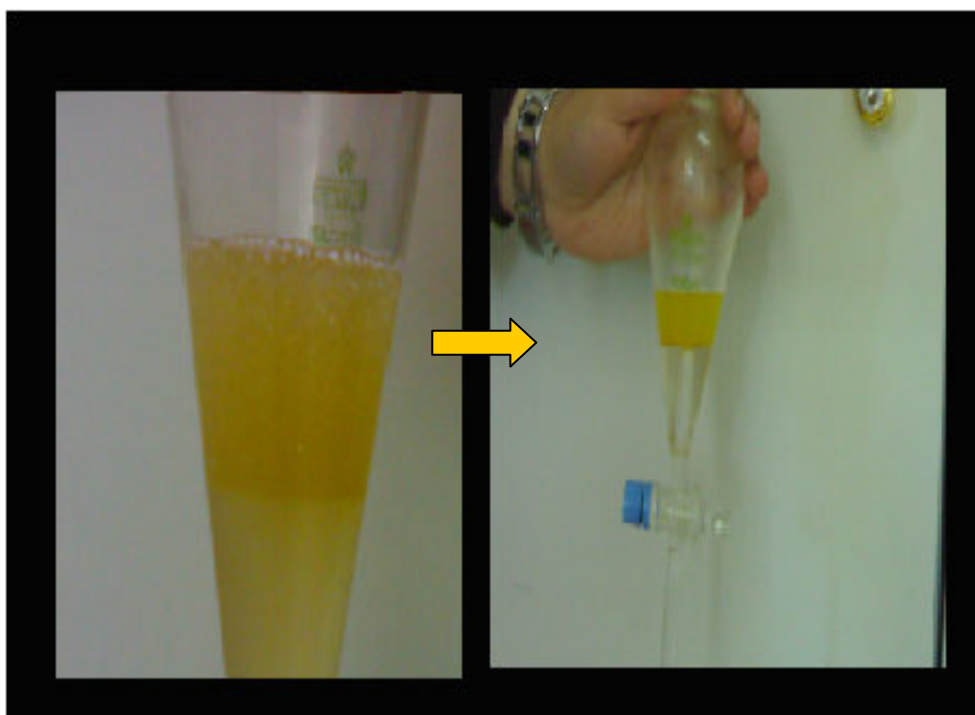


Figure (2.5.1) Extraction of β -carotene

2.6 Freeze Drying:

β -carotene crystallization was accomplished using freeze dryer equipment at 0.32 hpa pressure and -85°C temperature for four hours.

2.7 Spectrophotometer Analysis

Scanning of the absorption spectrum of the hexane extracted pigments present in the supernatant was recorded at the wavelengths 350 – 700 nm using a scanning spectrophotometer. β -carotene was estimated spectrophotometrically at wave lengths 454 and 480 nm according to pharmacopeia, and then the concentration was determined.

5. The absorbance of the final solution is then measured at 440 nm against a cyclo hexane blank, and a spectrophotometer plot was obtained.

2.8 Enzymatic Oxidation

It is a process by which the enzyme is used as a catalyst in the oxidation reaction, to allow a new pathway for the reaction to occur. The enzymatic oxidation reaction occurs in a different potential range depending on the material reacted and the product produced. The produced Beta Carotene was transferred into a one liter capacity bioreactor and was subjected to enzymatic oxidation under inert atmosphere and at 37°C , and then subjected to freeze drying for 24 hours, and the product was tretinoin (vitamin A). It was analyzed properly. The structure of vitamin A is shown in Figure (2.8.1).

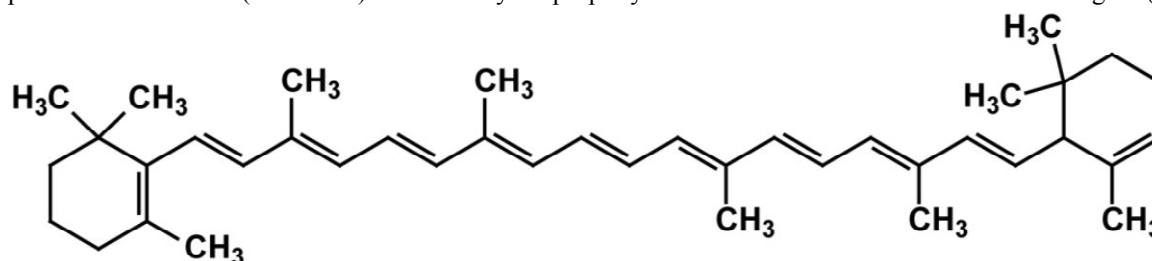


Fig.(2.8.1)The structure of vitamin A

Astaxanthin, the structure of which structure is shown in Figure (2.8.2) can be found in certain percentages in Beta Carotene and associated with vitamin A.

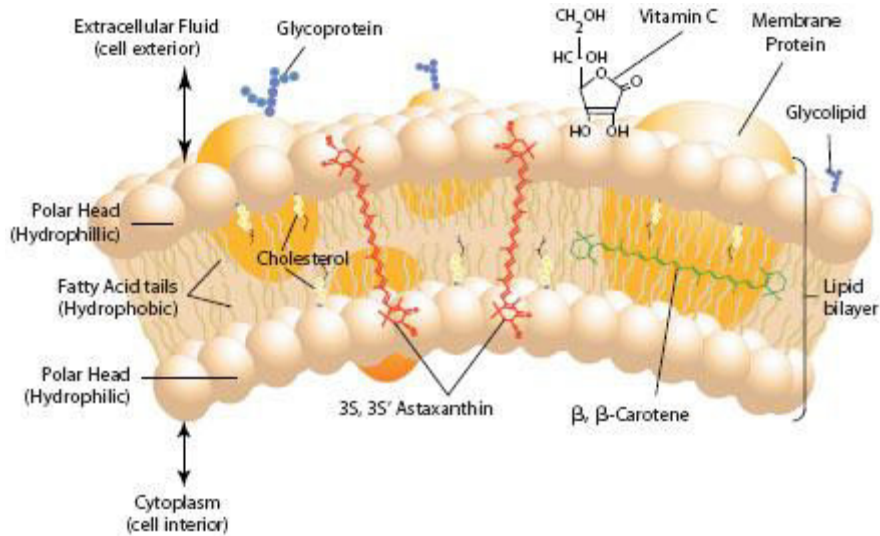


Fig.(2.8.2) Astaxanthin increases muscles endurance and strengthens the body's immune system.

2.9 Infrared Spectroscopy

Infrared spectroscopy (IR spectroscopy) is the spectroscopy that deals with the infrared region of the electromagnetic spectrum, that is light with a longer wavelength and lower frequency than visible light. It covers a range of techniques, mostly based on absorption spectroscopy]. IR spectrometers can accept a wide 64 range of sample types such as gases, liquids, and solids. The infrared portion of the electromagnetic spectrum is usually divided into three regions related to the visible spectrum:

- The near-infrared: the wavelength is approximately 0.8–2.5 μm .
- The mid-infrared: the wavelength is approximately 2.5–25 μm .
- The far-infrared: the wavelength is approximately 25–1000 μm .

3. Results and Discussions

a. Results:

3.1 Cultivation:

Dunaliella cells were found in samples collected from the Dead Sea. The cells were observed under digital microscope with USB camera.

The Dunaliella cells were cultivated in an inorganic medium (Johnson media); ordinary air was supplied at room temperature 25 ± 2 under high illumination.

3.2 Cell counting:

The number of Dunaliella cells was found to be about 500,000cell/ml using Hemocytometer slide under microscope .Cell filtration was performed for five samples and a sequential curve was plotted with slope near to 0.0407 (approximately horizontal line resulted) and an intercept of 2.1129 indicating a cell density of 2.1129 g dry weight per liter see Fig. (4.2.1).1 m³ cultivation media was found to contain approximately 114.1 g β -carotene.

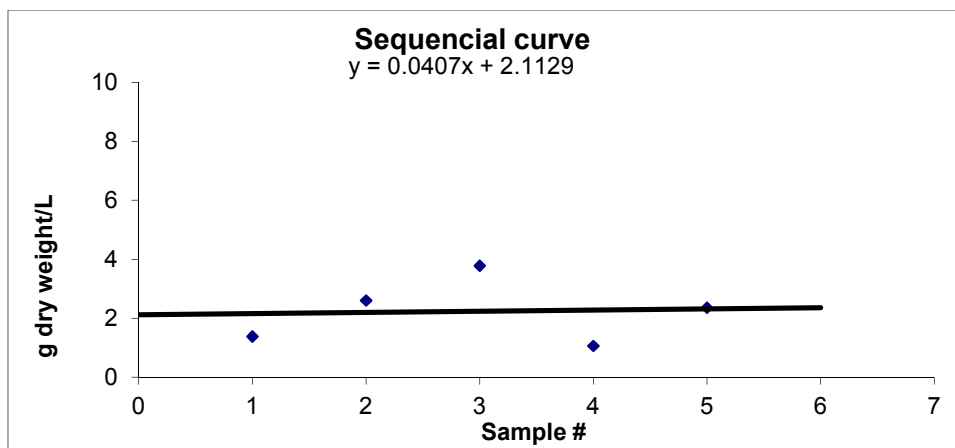


Fig (3.2.1): Sequential curve of filtered samples



Fig (3.3.1): β -carotene extracted with jojoba oil

3.4 Freeze Drying:

A centrifuged suspended in distilled water samples were dried using freeze dryer device at a temperature of $-85\text{ }^{\circ}\text{C}$ and 0.32 hpa pressure, the run time was four hours to obtain crystals. 1 g Beta Carotene powder was obtained.

3.5 Spectro photometric Analysis:

Scanning spectrum of the extracted pigments is shown in Fig. 4.5.1.

The highest peak was appeared at 440 nm wave length and indicated an optical density value near to 2.7 . The shorter peak indicates related carotenoids such as 9-cis, 13-cis, and 15-cis Astaxanthin.

b. Discussions:

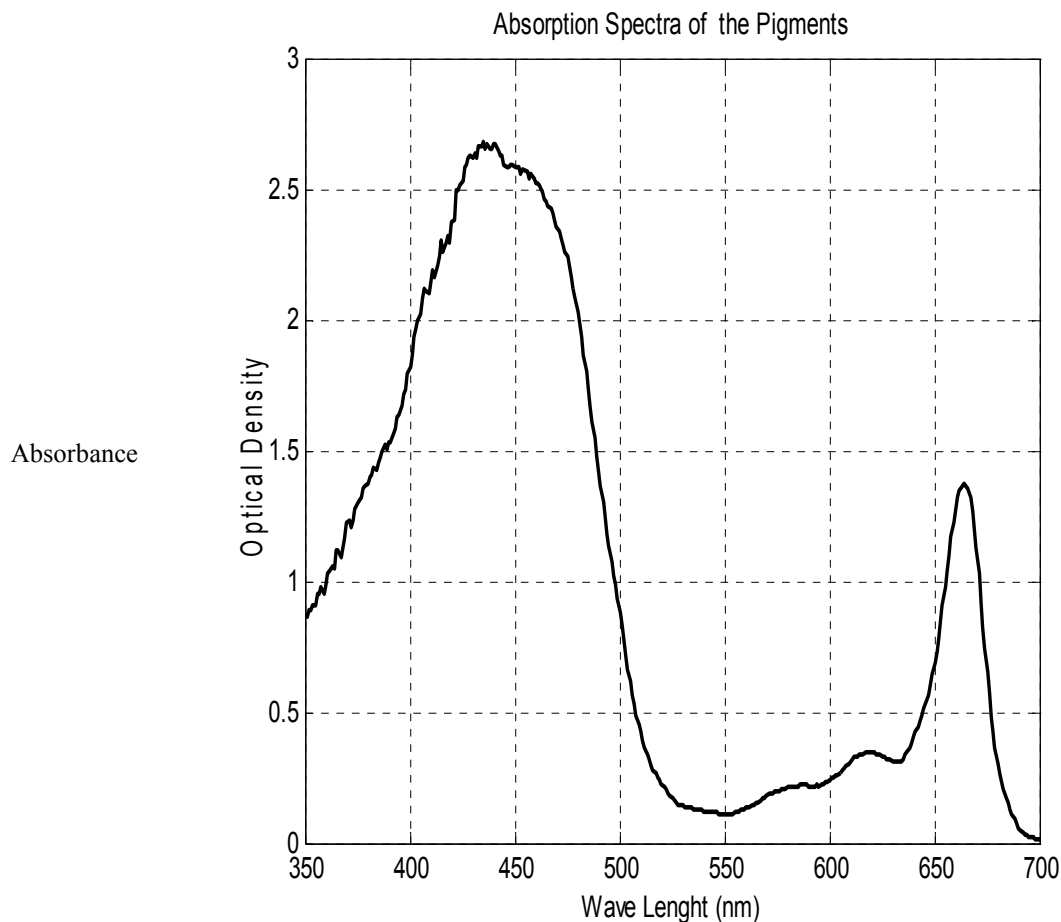


Fig. (3.5.1) Absorption spectra of Beta Carotene extracted with hexane.

Although β -carotene can be synthesized or extracted from other natural sources such as carrots, *Dunaliella* is still the richest and best natural source of this carotenoid. The most suitable media for the isolated *Dunaliella* species lab scale cultivation was found to be Modified Johnson Medium, which gave the highest

growth and β -carotene production. Pigment analysis using spectrophotometer indicates the presence of β -carotene, in addition to other related pigments. As can be seen from Figure (4.2.1) the dry weight per liter cultivation media was 2.1 gram of which 5.8% was β -carotene. Then in 1m^3 cultivation media 114.1 g highly pure β -carotene exists. To obtain 1 kg of highly purified β -carotene we need 8.8m^3 cultivation media, its price is about 5000 US\$/kg. The low cell densities achieved by the algae and their small cell size however, make harvesting more difficult and costly.

The best method for concentration the algal material without smashing its clusters was found to be gathering it in a separatory funnel and permit it to sediment by gravity force. The best extraction solvents were hexane, ether, jojoba oil or light vegetable oil for dietary use.

It was noted that sun-dried or heat dried samples of *Dunaliella* were rapidly degraded in terms of keeping β -carotene not oxidized, so freeze drying process was employed and showed high efficiency in drying samples without oxidizing any part of β -carotene.

Further more the use of β -carotene as a food or food additive and a nutritional supplement means that a high quality product is required. This means that great care must be taken in the extraction and formulation steps.

The extracted beta carotene was subjected to enzymatic oxidation using 15,15'-beta- beta carotene dioxygenase enzyme with a purity of 99.9 %, for one hour at 37 °C with and without nitrogen atmosphere.

The product was analyzed by using IR- Spectrophotometer and UV- Spectro photometry to detect the presence of beta-carotene and Vitamin A at a specific wave length. The wave number of IR spectrometer can be identified by the sharpest peak appears at the chromatogram, the wave number affected by many variables such as concentration, temperature. The appearance of more than one peak indicates the presence of carotenoids besides vitamin A.

After the enzymatic oxidation, the samples were analyzed by IR –Spectrophotometer, the results obtained at wave number 1636.19cm^{-1} and with the presences of broad peaks at the wave number $625.82, 2082.66, 3447.87\text{cm}^{-1}$. While pure Vitamin A have a wave number range ($1512.03 - 2920.02$) cm^{-1} . This indicated that the result obtained in the range of the standard Vitamin A. Figure (4.5.2) shows the produced natural vitamin A.



Figure (3.5.2) The produced natural vitamin A

4. Conclusions and Recommendations

4.1 Conclusions

1. *Dunaliella salina* was cultivated using a certain media and the cell count was performed to be 500,000 cells per ml after two weeks of cultivation.
2. Beta- carotene was extracted using n-hexane, jojoba oil and ethanol as a solvent.
3. UV- visible spectro photometric analysis was carried out for beta carotene.

4. The produced β -carotene was associated with related carotenoids such as 9-cis, 13-cis, and 15-cis Astaxanthin.
5. The enzymatic oxidation was carried out using a fermenter with and without nitrogen and with the enzyme 15,15'-beta- beta carotene dioxygenase.
6. The freeze drying step was performed at Jordan Bio- Industries Center (JOVAC).
7. The resulted vitamin A powder was analyzed using IR and UV-spectro photometric analysis at one of the local pharmaceutical companies.
8. The resulted natural vitamin A powder was within the expected limits as far as the quality and quantity analysis are concerned.
9. According to the obtained results a spin off company will be established with BASF/The Chemical Company/Germany, to produce natural vitamin A for the whole world.

4.2 Recommendations

1. Using a pilot plant of a fully controlled fermentor to carry out the enzymatic oxidation step.
2. Using a local cultivation media in the cultivation step.
3. Putting more emphasis on the purification steps.
4. Using two different enzymes to produce natural Astaxanthin from natural beta carotene.
5. A mixture of n-decane and $\text{CH}_2\text{CH}_2\text{Cl}$ is be used as a solvent in the extraction process.

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